

# One-Pot, Four-Component Synthesis of Fully Substituted 1,3,4-Oxadiazole Derivatives from (Isocyanoimino)triphenylphosphorane, a Primary Amine, an Aromatic Carboxylic Acid, and Chloroacetone

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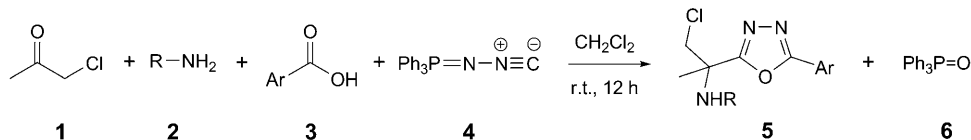
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The 1:1 imine intermediate **7** generated by the addition of a primary amine **2** to chloroacetone (**1**) is trapped by (isocyanoimino)triphenylphosphorane (**4**) in the presence of an aromatic carboxylic acid **3** and leads to the formation of the corresponding iminophosphorane intermediate **9** (Scheme 2). The 1,3,4-oxadiazole derivatives **5** are then formed *via* an intramolecular *aza-Wittig* reaction of the iminophosphorane intermediate **9**. The reactions were completed under neutral conditions at room temperature. The fully substituted 1,3,4-oxadiazole derivatives **5** were produced in high yields (Table).

**Introduction.** – In recent years, several synthetic methods have been reported for the preparation of (isocyanoimino)triphenylphosphorane (= *N*-(triphenylphosphorylidene)isocyanamide; CN–N=PPh<sub>3</sub>; **4**) [1][2]. There are several reports on the use of **4** in the synthesis of metal complexes [1][2]. However, application of **4** in the synthesis of organic compounds is fairly rare [3–13]. As part of our ongoing program to develop efficient and robust methods for the preparation of organic compounds [14–28], we sought to develop a convenient preparation of fully substituted 1,3,4-oxadiazole derivatives **5a–5q**. Herein, we report a four-component reaction, which, starting from readily available chloroacetone (**1**), affords fully substituted 1,3,4-oxadiazole derivatives **5a–5q** (Scheme 1).

Scheme 1. Four-Component Synthesis of 1,3,4-Oxadiazoles **5** (see Table)



**Results and Discussion.** – The 1:1 imine intermediate (= *N*-alkylideneamine intermediate) generated by the condensation reaction of chloroacetone (**1**) with primary amines **2** is trapped by **4** in the presence of aromatic carboxylic acids **3** and leads to the formation of 1,3,4-oxadiazole derivatives **5** and Ph<sub>3</sub>P=O (**6**; Scheme 1 and Table). The reaction proceeds smoothly and cleanly under mild and neutral conditions, and no side reactions were observed. The structures of the products were deduced from

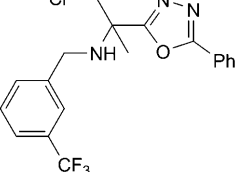
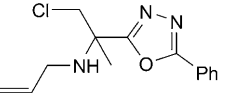
Table. *Synthesis of 2,5-Disubstituted 1,3,4-Oxadiazole Derivatives 5a–5q from Chloroacetone (1), Primary Amine 2, and Carboxylic Acid 3 in the Presence of (Isocyanoimino)triphenylphosphorane (4)*

R	Ar	Product	Yield [%] <sup>a)</sup>	M.p. [°]	IR <sup>b)</sup> [cm <sup>-1</sup> ]
Bn	Ph		<b>5a</b> 85	82–84	1608, 1449
Bn	4-Br–C <sub>6</sub> H <sub>4</sub>		<b>5b</b> 87	99–101	1602, 1482
Bn	4-MeO–C <sub>6</sub> H <sub>4</sub>		<b>5c</b> 77	oil	1616, 1496
Bn	4-Me–C <sub>6</sub> H <sub>4</sub>		<b>5d</b> 78	70–72	1618, 1499
Bn	4-Cl–C <sub>6</sub> H <sub>4</sub>		<b>5e</b> 75	80–82	1608, 1486
4-Methyl- benzyl	Ph		<b>5f</b> 76	71–73	1609, 1449
4-Methoxy- benzyl	4-Br–C <sub>6</sub> H <sub>4</sub>		<b>5g</b> 75	95–97	1605, 1478
Bn	4-NC–C <sub>6</sub> H <sub>4</sub>		<b>5h</b> 75	oil	1656, 1494

Table (cont.)

R	Ar	Product	Yield [%] <sup>a)</sup>	M.p. [°]	IR <sup>b)</sup> [cm <sup>-1</sup> ]
2-Chlorobenzyl	Ph		<b>5i</b> 82	65–67	1609, 1475
Bn	3,4-Me <sub>2</sub> C <sub>6</sub> H <sub>3</sub>		<b>5j</b> 70	oil	1617, 1455
Bn	4-F-C <sub>6</sub> H <sub>4</sub>		<b>5k</b> 80	77–79	1612, 1499
4-Fluorobenzyl	4-Me-C <sub>6</sub> H <sub>4</sub>		<b>5l</b> 70	oil	1614, 1499
Furan-2-ylmethyl	4-Cl-C <sub>6</sub> H <sub>4</sub>		<b>5m</b> 80	79–81	1609, 1485
Bn	2-Br-C <sub>6</sub> H <sub>4</sub>		<b>5n</b> 86	oil	1602, 1455
3,4-Dichlorobenzyl	Ph		<b>5o</b> 85	73–75	1612, 1471

Table (cont.)

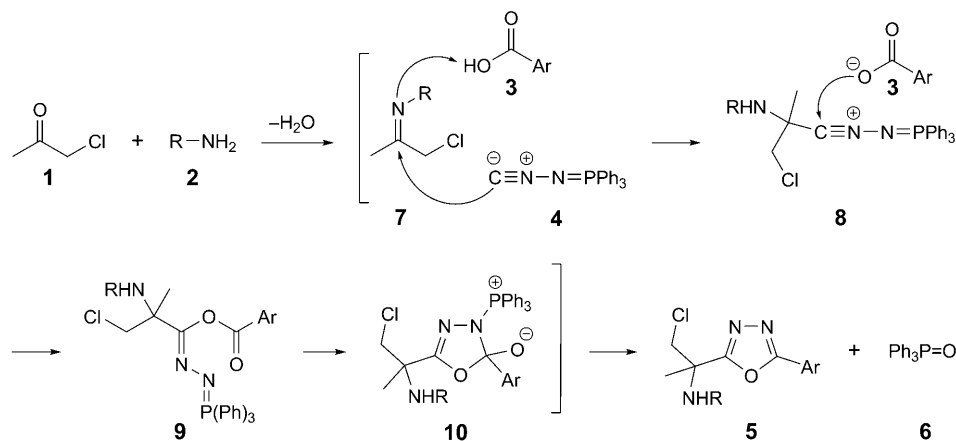
R	Ar	Product	Yield [%] <sup>a)</sup>	M.p. [°]	IR <sup>b)</sup> [cm <sup>-1</sup> ]	
3-(Trifluoromethyl)-benzyl	Ph		<b>5p</b>	77	oil	1614, 1452
Allyl	Ph		<b>5q</b>	70	oil	1649, 1451

<sup>a)</sup> Yield of isolated **5**. <sup>b)</sup> Two main absorptions in KBr.

their IR, <sup>1</sup>H- and <sup>13</sup>C-NMR, and mass spectra. For example, the <sup>1</sup>H-NMR spectrum of **5a** consisted of an *AB* pattern for the H-atoms of the CH<sub>2</sub>Cl group ( $\delta$ (H) 3.69 and 3.75 (<sup>2</sup>*J*(H,H) = 12.0 Hz)), and a second *AB* pattern for the H-atoms of the PhCH<sub>2</sub> group ( $\delta$ (H) 3.96 and 4.07 (<sup>2</sup>*J*(H,H) = 11.5 Hz)), a *s* for NH ( $\delta$ (H) 2.22), and a *s* for Me ( $\delta$ (H) 1.77). The Ar groups exhibited characteristic signals in the aromatic region of the spectrum. The <sup>1</sup>H-decoupled <sup>13</sup>C-NMR spectrum of **5a** showed 14 distinct signals; partial assignment of these signals is given in the *Exper. Part*. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of compounds **5b**–**5q** were similar to those of **5a**, except for the aromatic moieties, which exhibited characteristic signals with appropriate chemical shifts.

A mechanistic pathway for the reaction is provided in *Scheme 2*. On the basis of the chemistry of isocyanides, it is reasonable to assume that the first step may involve the

Scheme 2. Proposed Mechanism for the Formation of Sterically Congested 1,3,4-Oxadiazole Derivatives **5**



formation of imine **7** by the condensation reaction of chloroacetone (**1**) with the primary amine **2**. The next step may involve nucleophilic addition of (isocyanoimino)-triphenylphosphorane (**4**) to the imine intermediate **7**, which is facilitated by its protonation with the carboxylic acid **3**, leading to nitrilium intermediate **8**. This intermediate may be attacked by the conjugate base of the carboxylic acid to form the 1:1:1 adduct **9**. The intermediate **9** then undergoes an intramolecular aza-Wittig reaction [3–13] of the iminophosphorane moiety with the ester C=O group to afford the isolated sterically congested 1,3,4-oxadiazole derivative **5** by removal of  $\text{Ph}_3\text{P}=\text{O}$  (**6**) from intermediate **10**.

We also used 1,1-dichloroacetone, 1,3-dichloroacetone, 2-chloroacetophenone, 2-bromoacetophenone, trifluoroacetone, and 2,2-dichlorobutyrophenone instead of chloroacetone **1** in this reaction, but no corresponding products **5** were observed. In all the cases, several colored products were detected by TLC monitoring.

**Conclusions.** – We believe that the reported method offers a mild, simple, and efficient route for the preparation of fully substituted 1,3,4-oxadiazol derivatives of type **5**. Ease of workup, high yields, and fairly mild reaction conditions make it a useful addition to modern synthetic methodologies. Other aspects of this synthetic process are under investigation.

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### Experimental Part

**General.** (Isocyanoimino)triphenylphosphorane (= *N*-triphenylphosphoranylidene)isocyanamide; **4**) was prepared based on reported procedures [1][2]. Other starting materials and solvents were obtained from *Merck* (Germany) and *Fluka* (Switzerland) and were used without further purification. Reaction monitoring by TLC and NMR (no by-products formed). Prep. TLC: plates coated with *Merck* silica gel  $F_{254}$  powder. M.p.: *Electrothermal-9100* apparatus; uncorrected. IR Spectra: *Jasco-6300* FT-IR spectrometer;  $\tilde{\nu}$  in  $\text{cm}^{-1}$ .  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR Spectra ( $\text{CDCl}_3$ ): *Bruker-DRX-250-Avance* spectrometer; at 250.0 and 62.5 MHz, resp.; in  $\text{CDCl}_3$ ;  $\delta$  in ppm rel. to  $\text{Me}_4\text{Si}$  as internal standard,  $J$  in Hz. MS: *Finnigan-Matt-8430* mass spectrometer operating at an ionization potential of 20 eV; in  $m/z$  (rel. %). Elemental analyses: *Heraeus CHN-O-Rapid* analyzer.

**General Procedure.** To a magnetically stirred soln. of primary amine **2** (1 mmol), chloroacetone (**1**; 1 mmol), and (isocyanoimino)triphenylphosphorane (**4**; 1 mmol) in  $\text{CH}_2\text{Cl}_2$  (5 ml) was added dropwise a soln. of benzoic acid derivative **3** (1 mmol) in  $\text{CH}_2\text{Cl}_2$  (5 ml) at r.t. over 15 min. The mixture was stirred for 12 h. The solvent was evaporated and the viscous residue purified by prep. TLC (silica gel  $F_{254}$ , petroleum ether/AcOEt 4:1). Purified **5a–5q** were obtained as yellow powders or yellow oils. For yields, m.p., and IR of **5a–5q**, see Table<sup>1</sup>).

***N*-Benzyl-*N*-[2-chloro-1-methyl-1-(5-phenyl-1,3,4-oxadiazol-2-yl)ethyl]amine** (=  $\alpha$ -(Chloromethyl)- $\alpha$ -methyl-5-phenyl-*N*-(phenylmethyl)-1,3,4-oxadiazol-2-methanamine; **5a**): Yellow powder. IR (KBr): 3456, 3269, 2927, 1608, 1449, 1069.  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ ): 1.77 (s, Me); 2.23 (s, NH); 3.69, 3.76 (AB,  $^2J = 12.3$ ,  $\text{CH}_2\text{Cl}$ ); 3.96, 4.07 (AB,  $^2J = 11.5$ ,  $\text{PhCH}_2$ ); 7.24–8.07 (m, 10 arom H).  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ ): 22.53 (Me); 47.82, 50.24. (2  $\text{CH}_2$ ); 57.58 (CNH); 127.00, 127.31, 128.27, 128.51, 129.07, 131.8 (10 CH); 123.70, 139.33 (2 C); 165.24, 167.77 (2 C=N). EI-MS: 327 ( $M^+$ ), 187 (16), 149 (33), 105 (70), 91 (100), 76 (29), 43 (54). Anal. calc. for  $\text{C}_{18}\text{H}_{18}\text{ClN}_3\text{O}$  (327.81): C 65.95, H 5.53, N 12.82; found: C 65.83, H 5.49 N 12.77.

<sup>1</sup>) Supplementary material (characterization data of **5b–5q**) may be obtained upon request from the senior author A. R

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